This article was downloaded by:

On: 28 January 2011

Access details: Access Details: Free Access

Publisher Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.informaworld.com/smpp/title~content=t713618290

Bioorganogermanium Chemistry: Studies on C/Si/Ge Bioisosterism

R. Tacke; T. Heinrich; T. Kornek; M. Merget; S. A. Wagner; J. Gross; C. Keim; G. Lambrecht; E. Mutschler; T. Beckerss; M. Bernd; T. Reissmann

To cite this Article Tacke, R. , Heinrich, T. , Kornek, T. , Merget, M. , Wagner, S. A. , Gross, J. , Keim, C. , Lambrecht, G. , Mutschler, E. , Beckerss, T. , Bernd, M. and Reissmann, T.(1999) 'Bioorganogermanium Chemistry: Studies on C/Si/Ge Bioisosterism', Phosphorus, Sulfur, and Silicon and the Related Elements, 150: 1, 69-87

To link to this Article: DOI: 10.1080/10426509908546372 URL: http://dx.doi.org/10.1080/10426509908546372

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.informaworld.com/terms-and-conditions-of-access.pdf

This article may be used for research, teaching and private study purposes. Any substantial or systematic reproduction, re-distribution, re-selling, loan or sub-licensing, systematic supply or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Bioorganogermanium Chemistry: Studies on C/Si/Ge Bioisosterism

R. TACKE^a, T. HEINRICH^a, T. KORNEK^a, M. MERGET^a, S. A. WAGNER^a, J. GROSS^b, C. KEIM^b, G. LAMBRECHT^b, E. MUTSCHLER^b, T. BECKERS^c, M. BERND^c and T. REISSMANN^c

^aInstitut für Anorganische Chemie, Universität Würzburg, Am Hubland, D-97074 Würzburg, Germany, ^bPharmakologisches Institut für Naturwissenschaftler, Biozentrum Niederursel, Universität Frankfurt, Marie-Curie-Straße 9, Geb. N260, D-60439 Frankfurt, Germany and ^cASTA Medica AG, Weismüllerstraße 45, D-60314 Frankfurt, Germany

In context with systematic investigations on C/Si/Ge bioisosterism, the following studies were carried out: (a) synthesis and pharmacological characterization of centrochiral enantiomerically pure germanium-based muscarinic antagonists; (b) synthesis and pharmacological characterization of a germanium-containing decapeptide; (c) studies on the metabolism of a germanium-based drug in the rat; (d) synthesis of centrochiral enantiomerically pure germanes using biotransformations with whole microorganisms or isolated enzymes. These investigations demonstrated that there are distinct bioisosteric relationships between the C/Si/Ge analogues studied.

Keywords: bioorganogermanium chemistry; C/Si/Ge bioisosterism; sila-and germa-drugs; silicon- and germanium-containing α -amino acids and peptides; metabolism of sila- and germa-drugs; stereoselective biocatalysis

INTRODUCTION

Bioorganosilicon chemistry represents a fascinating, rapidly expanding branch of organosilicon chemistry (for reviews, see refs. 1–8). In recent years, practical aspects of this research field have become of increasing importance. The development of silicon-based drugs and agrochemicals and the application of biocatalysis in synthetic organosilicon chemistry are examples of this. Compared to the extensive research activities in bioorganosilicon chemistry, bioorganogermanium chemistry is significantly less explored (for reviews, see refs. 9 and 10). In this article we report on some results of our own studies in this particular field. The investigations presented here were carried out with a special emphasis on the aspect C/Si/Ge bioisosterism.

CHIRAL GERMANIUM-BASED MUSCARINIC ANTAGONISTS

During the past decade, we have developed a variety of highly potent and receptor-selective silicon-based muscarinic antagonists (for selected publications, see refs. 11–15 and references cited therein). Compound

(R)-2b is an example of this particular type of drug. [11]

In the course of these studies, we became also interested in the muscarinic receptor binding of related carbon and germanium analogues, such as compounds (R)-2a^[15,16] and (R)-2c^[14]. We report here on the syntheses and antimuscarinic properties of the enantiomerically pure (R)- and (S)-enantiomers of the centrochiral C/Si/Ge analogues 1a/1b/1c, 2a/2b/2c, 3a/3b/3c, and 4a/4b/4c.

CI
$$CH_2CI$$
 CH_2CI CH_2CI

SCHEME 1

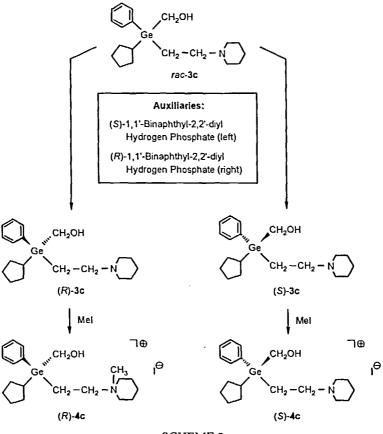
The strategy for the synthesis of the racemic germanes 1c^[14] and 3c^[17] is

outlined in Scheme 1 (example: synthesis of *rac-3c*). The corresponding silicon analogues *rac-1b*^[11] and *rac-3b*^[17] were prepared analogously [starting from (MeO)₃SiCH₂Cl], whereas the carbon compounds *rac-1a*^[16] and *rac-3a*^[17] were obtained by a quite different approach as shown in Scheme 2 (example: synthesis of *rac-3a*).

SCHEME 2

As demonstrated in Scheme 3, the enantiopure compounds (R)-3c and (S)-3c were obtained by classical resolution of rac-3c using the antipodes of 1,1'-binaphthyl-2,2'-diyl hydrogen phosphate as resolving agents. Subsequent quaternization of (R)-3c and (S)-3c with methyl iodide gave the enantiopure compounds (R)-4c and (S)-4c, respectively. The (R)- and (S)-enantiomers of 1a, (S)-11 1c, (S)-12 2a, (S)-13 2c, (S)-14 3a, (S)-17 3b, (S)-17 and 4b, were prepared analogously. For these syntheses, the

antipodes of 1,1'-binaphthyl-2,2'-diyl hydrogen phosphate (for 1a), 2,3-di-p-toluoyltartaric acid (for 1b, 1c, and 3b), and 2,3-dibenzoyltartaric acid (for 3a) were used as resolving agents.



SCHEME 3

The (R)- and (S)-enantiomers of the C/Si/Ge analogues 1a/1b/1c and 2a/2b/2c were studied in functional pharmacological experiments for their affinities (pA₂ values) at muscarinic M1 (rabbit vas deferens), M2 (guineapig atria), and M3 receptors (guineapig ileum) using 4-F-PyMcN' (M1)

or arecaidine propargyl ester (M2, M3) as the agonist. The antipodes of the C/Si/Ge analogues 3a/3b/3c and 4a/4b/4c were investigated for their affinities (p K_i values) at recombinant human muscarinic receptor subtypes m1-m5 stably expressed in CHO cells (binding studies with $[^3H]$ -N-methylscopolamine as the radioligand). Generally, very similar stereoselectivities and pharmacological selectivities were observed in these studies for the respective C/Si/Ge analogues, indicating strongly pronounced bioisosteric relationships between these analogous carbon, silicon, and germanium compounds. This is illustrated for the antipodes of 2a/2b/2c and 4a/4b/4c in Figures 1 and 2.

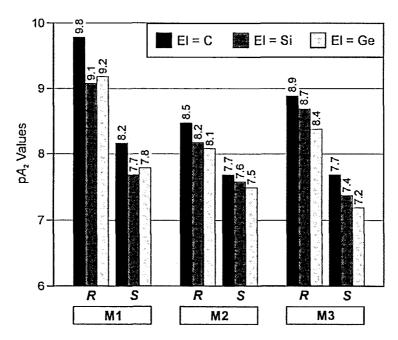


FIGURE 1 Affinity profiles of the antipodes of the C/Si/Ge analogues 2a/2b/2c at muscarinic M1, M2, and M3 receptors.

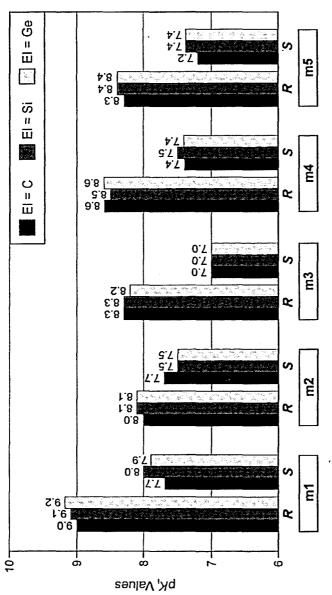


FIGURE 2 Affinity profiles of the antipodes of the C/Si/Ge analogues 4a/4b/4c at human muscarinic m1, m2, m3, m4, and m5 receptors.

GERMANIUM-CONTAINING α-AMINO ACIDS AND A GERMANIUM-CONTAINING DECAPEPTIDE

Synthetic amino acids with unnatural side chains have proven useful for probing the structural requirements for the biological activity of numerous peptides and proteins. In addition, unnatural amino acids are of interest as precursors of drugs and plant-protective agents. In context with our studies on silicon-containing α -amino acids, we have succeeded in synthesizing the first germanium-containing α -amino acids, L- and D-(trimethylgermyl)-alanine [L-5c and D-5c]. [18]

OH

$$EIMe_3$$
 $L-5a/L-5b/L-5c$

a: $EI = C$
b: $EI = Si$
c: $EI = Ge$

$$D-5a/D-5b/D-5c$$

These compounds and their corresponding carbon and silicon analogues, the α -amino acids L-5a, D-5a, L-5b, and D-5b, are interesting building blocks for new biologically active peptides.

We report here on the first germanium-containing peptide, the decapeptide 6c, which is a derivative of the GnRH (gonadotropin-releasing hormone) antagonist Cetrorelix^{INN} (replacement of L-tyrosine by L-5c in position 5; for recent reviews dealing with Cetrorelix, see refs. 19 and 20). The biological properties of the germanium-containing decapeptide 6c were compared with those of its corresponding carbon (6a) and silicon analogue (6b). [21]

SCHEME 4

The α -amino acids L-5c and D-5c were prepared according to Scheme 4 by three-step syntheses and isolated as almost enantiomerically pure crystalline products. The decapeptide 6c was prepared by solid-phase synthesis (SPPS) using the Fmoc-protected α -amino acid, compound L-7.

The decapeptides 6a and 6b were synthesized analogously. The identity of 6a, 6b, and 6c was established by NMR studies and mass-spectrometric investigations. As an example of this, an electrospray mass spectrum of the germanium-containing decapeptide 6c is shown in Figure 3.

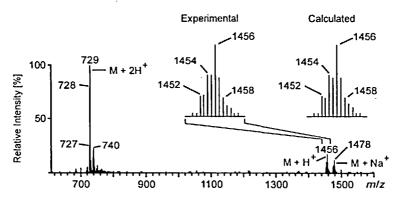


FIGURE 3 Electrospray mass spectrum of the germanium-containing decapeptide 6c.

The C/Si/Ge analogues 6a/6b/6c were studied for their antagonistic potencies at the human GnRH receptor using a functional reporter gene assay (tiptorelin as the agonist). The IC₅₀ values [nM] determined were as

follows: 0.93 (6a), 0.75 (6b), 0.50 (6c). These preliminary data demonstrate that all decapeptides are potent GnRH antagonists that differ only slightly in their antagonistic potencies, the germanium compound 6c being somewhat more potent than its carbon and silicon analogue. Thus, there are distinct bioisosteric relationships between the C/Si/Ge analogues 6a/6b/6c.

Compounds 6a, 6b, and 6c were also studied for their in vivo properties in the castrated male rat after s.c. administration (0.05 mg/kg). As illustrated for the germanium-containing decapeptide 6c in Figure 4, a serum LH (luteinizing hormone) suppression was observed after administration of the C/Si/Ge analogues 6a/6b/6c. Again, distinct bioisosteric relationships were observed for these compounds. Interestingly, the decapeptides 6b and 6c were found to act ca. 6 times longer than their carbon analogue 6a.

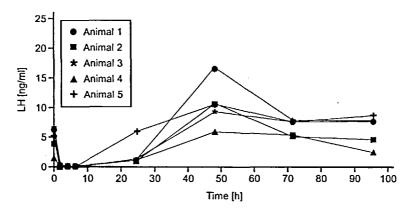


FIGURE 4 Serum LH suppression in the castrated male rat after s.c. administration of the germanium-containing decapeptide 6c (0.05 mg/kg).

METABOLISM OF A GERMANIUM-BASED MUSCARINIC ANTAGONIST IN THE RAT

The C/Si/Ge analogues 8a/8b/8c were studied for their metabolic fate in the rat. [22]

$$CH_2OH$$
 EI
 CH_2-CH_2-N
 $CH_$

The phase-I metabolism of these muscarinic antagonists was investigated after p.o. administration of 80 mg/kg. For this purpose, urine samples (collected for a period of 24 hours after administration) were studied with

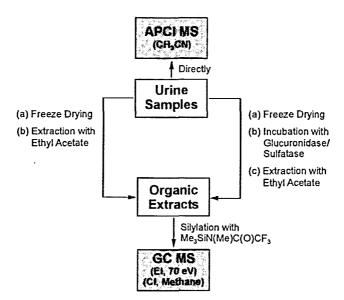


FIGURE 5 Experimental design used for the investigation of the phase-I metabolism of the C/Si/Ge analogues 8a/8b/8c in the rat after p.o. administration.

mass-spectrometric techniques (for the experimental design, see Figure 5). As shown in Figure 6, APCI MS studies of the untreated urine samples demonstrated the presence of the unchanged parent drugs 8a/8b/8c (peaks A) and three types of metabolites (peaks B, C, and D).

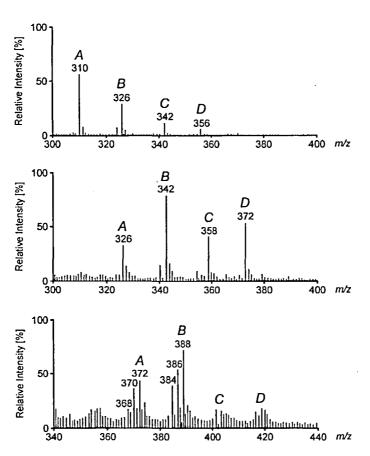


FIGURE 6 APCI MS spectra of urine samples collected after p.o. administration of 80 mg/kg of 8a (above), 8b (middle), and 8c (below).

To get information about the structure of these metabolites, organic extracts of the freeze-dried urine samples were silylated with Me₃SiN(Me)C(O)CF₃ and subsequently studied by GC MS experiments (EI, 70 eV; CI, methane). These investigations included urine samples which were treated with glucuronidase/sulfatase (see Figure 5) in order to cleave potential conjugates. The mass-spectrometric studies clearly demonstrated the presence of the metabolites 9a/9b/9c, 10a/10b/10c, and 11a/11b/11c. These preliminary results indicate a similar phase-I metabolism of the C/Si/Ge analogues 8a/8b/8c.

HO
$$CH_2OH$$
 CH_2OH
 CH_2OH
 CH_2-CH_2-N
 $9a/9b/9c$
 CH_2-CH_2-N
 OH_2OH
 OH_2OH

BIOCATALYSIS AS A PREPARATIVE METHOD FOR THE SYNTHESIS OF OPTICALLY ACTIVE CENTROCHIRAL GERMANES

By analogy with many organic ketones of the general formula type R₃C-C(O)-CR₃, related silaketones (acylsilanes) R₃Si-C(O)-CR₃ were found to be accepted as substrates by a variety of ketone-reducing microorganisms and to be converted stereoselectively into the corresponding optically active reduction products R₃Si-C(OH)H-CR₃ (see ref.

8 and literature cited therein). Related germaketones (acylgermanes) R₃Ge-C(O)-CR₃ were also found to undergo stereoselective microbial reductions, indicating again bioisosteric relationships between analogous carbon, silicon, and germanium compounds. The enantioselective microbial reductions shown in Schemes 5-7 are examples of this.^[23]

The acetylgermane rac-12 was found to be reduced (R)-selectively by resting free cells of Saccharomyces cerevisiae (DHW S-3) to give a mixture of the diastereomeric (1-hydroxyethyl)germanes (GeR,CR)-13 and (GeS,CR)-13 (molar ratio 1:1) (Scheme 5). These compounds could be separated by column chromatography on silica gel as almost diastereomerically and enantiomerically pure products.

SCHEME 5

As shown in Scheme 6, the related acetyldigermane *rac*-14 could also be reduced (R)-selectively with resting free cells of *Saccharomyces cerevisiae* (DHW S-3) to give the diastereomeric (1-hydroxyethyl)digermanes (GeR,CR)-15 and (GeS,CR)-15 (molar ratio 1:1).

SCHEME 6

The enantiomeric purities of these products were >98% ee. Interestingly, the Ge-Ge bonds of the substrate and the products are rather stable against hydrolytic cleavage under the bioconversion conditions used.

SCHEME 7

A quite different result was obtained when replacing the Me₃Ge group of rac-14 by a Me₃Si moiety (\rightarrow rac-16): Incubation of the acetyl(trimethylsilyl)germane rac-16 with resting free cells of Saccharomyces cerevisiae (DHW S-3) gave a 1:1 mixture of the diastereomeric (1-hydroxyethyl)hydridogermanes (GeS;CR)-17 (\geq 98% ee) and (GeR,CR)-17 (\geq 98% ee); i.e. a (probably chemically induced) Ge-Si cleavage was observed. The biotransformations outlined in Schemes 5-7 were performed on a preparative scale indicating that biocatalysis is an efficient preparative method for the synthesis of optically active germanes and digermanes.

SCHEME 8

As illustrated in Scheme 8, bioconversions with isolated enzymes have also a high potential for synthetic germanium chemistry. [24] The optically active hydridogermane (R)-20 (86% ee) was prepared on a preparative scale by an enantioselective transesterification of the prochiral diol 18 with ethyl acetate (acyl donor and reaction medium) using porcine pancreas lipase (PPL, E.C. 3.1.1.3) as the biocatalyst. The corresponding antipode (S)-20 (94% ee) was obtained by a PPL-catalyzed hydrolysis of the prochiral diacetate 19 (phosphate buffer/tetrahydrofuran as reaction medium). Analogous enzymatic conversions of related diols $R^1R^2EI(CH_2OH)_2$ (El = C, Si) and diacetates $R^1R^2EI(CH_2OAc)_2$ (El = C, Si) with lipases are also known (see ref. 24 and literature cited therein).

CONCLUDING REMARKS

The respective C/Si/Ge analogues described in this article were found to undergo quite similar interactions with biological systems, indicating that there are distinct bioisosteric relationships between analogous carbon, silicon, and germanium compounds.

Acknowledgements

R. T. wishes to express his sincere thanks to his coworkers and colleagues without whose contributions this article could not have been written; their names are cited in the references. In addition, financial support of our work by the *Deutsche Forschungsgemeinschaft* and the *Fonds der Chemischen Industrie* and support by the ASTA Medica AG (Frankfurt, Germany), the *Bayer AG* (Leverkusen and Wuppertal-Elberfeld, Germany), and the Merck *KGaA* (Darmstadt, Germany) is gratefully acknowledged.

References

- [1] R. J. Fessenden and J. S. Fessenden, Adv. Drug Res., 4, 95 (1967).
- [2] R. Tacke and U. Wannagat, Top. Curr. Chem., 84, 1 (1979).
- [3] M. G. Voronkov, Top. Curr. Chem., 84, 77 (1979).
- [4] R. J. Fessenden and J. S. Fessenden, Adv. Organomet. Chem., 18, 275 (1980).
- [5] R. Tacke and H. Zilch, Endeavour, New Series, 10, 191 (1986).
- [6] R. Tacke and B. Becker, Main Group Met. Chem.. 10, 169 (1987).
- [7] R. Tacke and H. Linoh, in S. Patai and Z. Rappoport, eds "The Chemistry of Organic Silicon Compounds, Part 2", Wiley, Chichester, 1989, pp. 1143–1206.
- [8] R. Tacke and S. A. Wagner, in Z. Rappoport and Y. Apeloig, eds., "The Chemistry of Organic Silicon Compounds, Part 3", Vol. 2, Wiley, Chichester, 1998, pp. 2363–2400.
- [9] E. Lukevics and L. Ignatovich, Appl. Organomet. Chem., 6, 113 (1992).
- [10] E. Lukevics, S. Germane, and L. Ignatovich, Appl. Organomet. Chem., 6, 543 (1992).

- [11] R. Tacke, D. Reichel, M. Kropfgans, P. G. Jones, E. Mutschler, J. Gross, X. Hou, M. Waelbreock, and G. Lambrecht, Organometallics 14, 251 (1995).
- [12] R. Tacke, D. Terunuma, A. Tafel, M. Mühleisen, B. Forth, M. Waelbroeck, J. Gross, E. Mutschler, T. Friebe, and G. Lambrecht, J. Organomet. Chem., 501, 145 (1995).
- [13] R. Tacke, B. Forth, M. Waelbroeck, J. Gross, E. Mutschler, and G. Lambrecht, J. Organomet. Chem., 505, 73 (1995).
- [14] R. Tacke, D. Reichel, P. G. Jones, X. Hou, M. Waelbroeck, J. Gross, E. Mutschler, and G. Lambrecht, J. Organomet. Chem., 521, 305 (1996).
- [15] E. Mutschler, H. A. Ensinger, J. Gross, A. Leis, K. Mendla, U. Moser, O. Pfaff, D. Reichel, K. Rühlmann, R. Tacke, M. Waelbroeck, J. Wehrle, and G. Lambrecht, *Pharmacochem. Libr.*, 24 (Perspectives in Receptor Research), 51 (1996).
- [16] J. Gross, G. Lambrecht, A. Leis, E. Mutschler, D. Reichel, K. Rühlmann, and R. Tacke, unpublished results.
- [17] T. Heinrich, C. Keim, T. Kornek, G. Lambrecht, E. Mutschler, and R. Tacke, unpublished results.
- [18] M. Merget, S. Bartoschek, R. Willeke, and R. Tacke, in N. Auner and J. Weis, eds., "Organosilicon Chemistry IV - From Molecules to Materials", Wiley-VCH, Weinheim, in press.
- [19] B. Kutscher, M. Bernd, T. Beckers, E. E. Polymeropoulos, and J. Engel, Angew. Chem., 109, 2240 (1997); Angew. Chem., Int. Ed. Engl., 36, 2148 (1997).
- [20] T. Reissmann, J. Engel, B. Kutscher, M. Bernd, P. Hilgard, M. Peukert, I. Szelenyi, S. Reichert, D. Gonzales-Barcena, E. Nieschiag, A. M. Comaru-Schally, and A V. Schally, *Drugs of the Future*, 19, 228 (1994).
- [21] T. Beckers, M. Bernd, M. Merget, T. Reissmann, and R. Tacke, unpublished results.
- [22] W. Dekant, E. Rosner, R. Tacke, and S. A. Wagner, unpublished results.
- [23] S. A. Wagner, S. Brakmann, and R. Tacke, in N. Auner and J. Weis, eds., "Organosilicon Chemistry II – From Molecules to Materials", VCH, Weinheim, 1996, pp. 237– 242.
- [24] R. Tacke, U. Kosub, S. A. Wagner, R. Bertermann, S. Schwarz, S. Merget, and K. Günther, Organometallics, 17, 1687 (1998).